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(12) United States Patent

Wu et al.

(54) SINTERING-ASSISTED DEPOSITION OF UNIFORM TITANIA NANOCRYSTALLINE COATINGS OVER AL FLAKES IN AQUEOUS SOLUTION

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U.S.C. 154(b) by 0 days.

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C09C 1/00 (2006.01)

805D 7/00 (2006.01)

(52) U.S. Cl.

(10) Patent No.:

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Jun. 7, 2016

67/0033; B82Y 20/00; B82Y 30/00; B82Y 40/00; B41M 5/52 See application file for complete search history.

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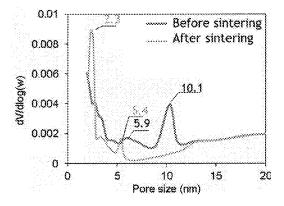
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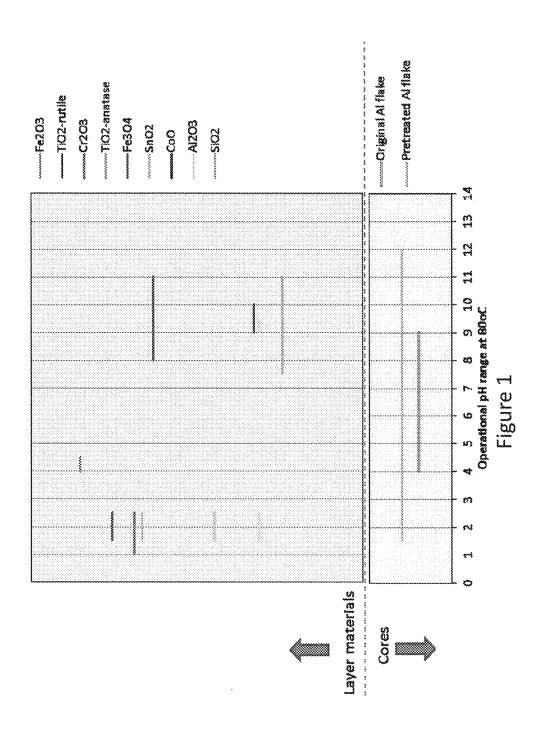
(57) ABSTRACT

A process of forming a multi-layered pigment comprising the steps of: providing a metal core material; treating the metal core material with an acid, depositing a passivation layer onto the metal core material; densifying the metal core material having the passivation layer reducing a pore size of the passivation layer; and depositing a high refractive index material onto the sintered material wherein the high refractive index layer is uniform and crack free.

19 Claims, 6 Drawing Sheets

Pore size distribution (BJH method)





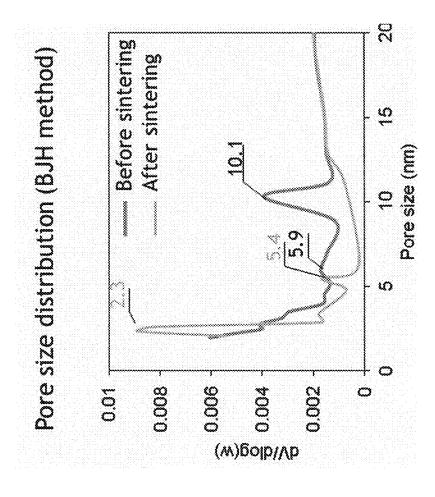


Figure 2

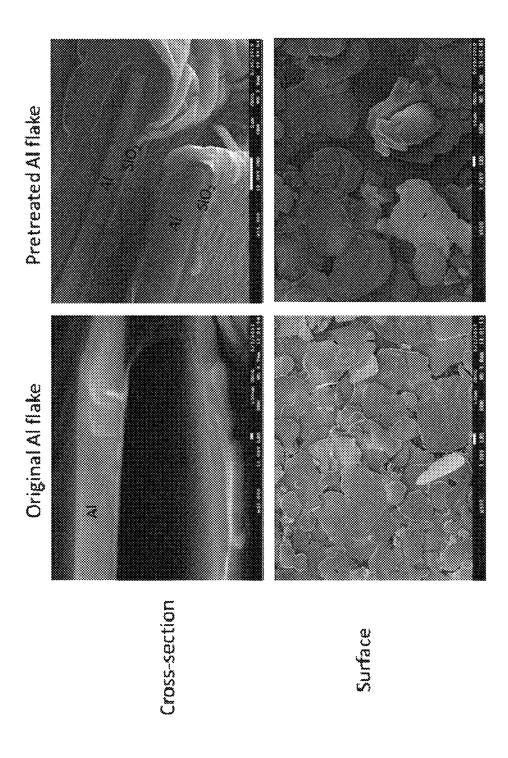


Figure 3

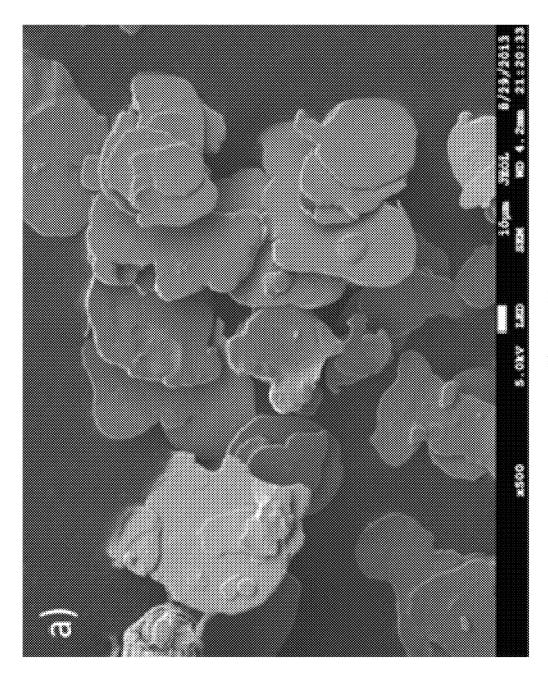


Figure 4A

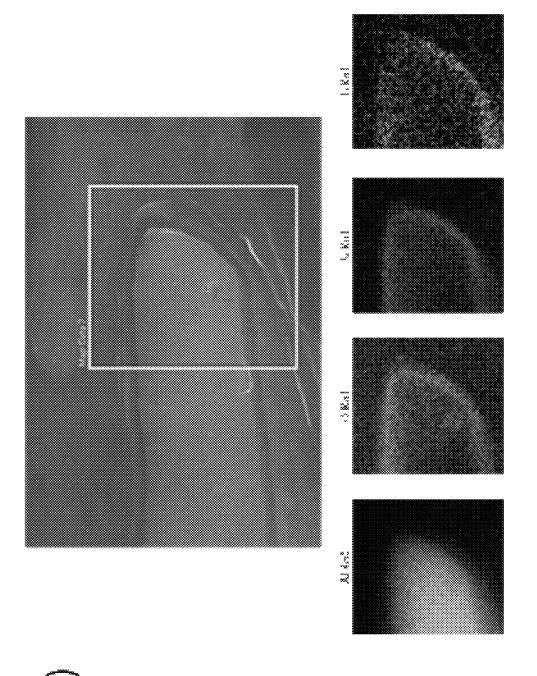


Figure 4B

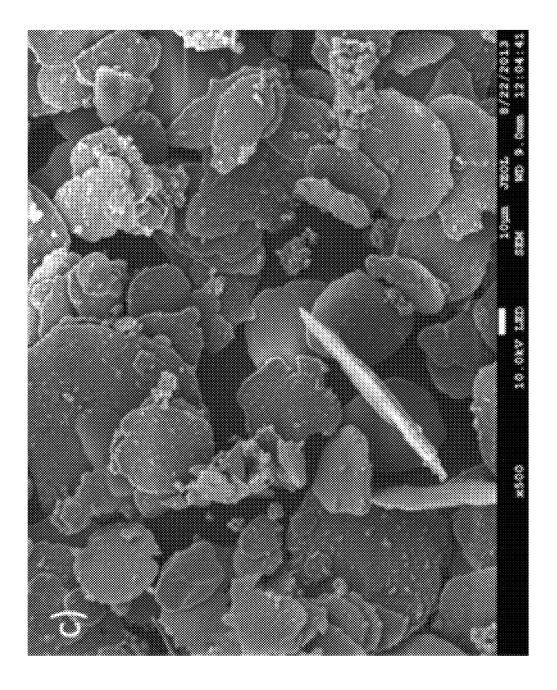


Figure 4C

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SINTERING-ASSISTED DEPOSITION OF UNIFORM TITANIA NANOCRYSTALLINE COATINGS OVER AL FLAKES IN AQUEOUS SOLUTION

FIELD OF THE INVENTION

The invention relates to processes for forming pigments having metal cores and passivation layers and a uniform smooth layer of a high refractive index material and materials formed by the process.

BACKGROUND OF THE INVENTION

Generally, luster pigments are known in the art for use in various applications such as automotive finishes, coatings and other pigment applications.

Such luster pigments may be formed by deposition of titanium oxide over highly reflective platelet-like aluminum 20 coated over a densifyed SiO₂—Al; flakes in a water system. The Al flake may include a protection layer of SiO₂. TiO₂ may be deposited to the Al flake under highly acidic conditions such as at a pH of less than 2.0 such that a hydrolysis reaction for TiO₂ may be achieved. However, such a process results in unsatisfactory coatings due to the 25 particle without densification showing a rough surface. diffusion of aqueous solution through the SiO₂ layer.

Problems associated with the above deposition include both etching of the Al core and a change in pH near the SiO₂—Al surface which are undesired for TiO₂ deposition. At such low pH, the protons in the aqueous solution can still 30 diffuse through the SiO₂ layer and react with the Al core during the typical long deposition period. This side reaction between the proton and Al as well as the resultant pH increase at the core surface renders the deposition of TiO₂ difficult. Diffusion of protons through the SiO₂ layer may cause a 35 reaction with Al such that hydrogen gas is released causing weak adhesion of the TiO₂ particles and the formation of channels or cracks in the SiO₂ and TiO₂ layers. Additionally, an increase in the pH may cause rapid deposition of the TiO₂ layer and formation of large particles of TiO2 which would 40 adversely affect the pigment's properties.

Further increasing the thickness of the protection layer of SiO₂ is also undesired as the low refractive index SiO₂ would reduce the overall color performance of pigments. There is therefore a need in the art for an improved process and pig- 45 ment that solves the problems identified above and produces a pigment that has a crack free and uniform TiO₂ layer.

SUMMARY OF THE INVENTION

In one aspect, there is disclosed a process of forming a multi-layered pigment comprising the steps of: providing a metal core material; treating the metal core material with an acid, depositing a passivation layer onto the metal core material; densifying the metal core material having the passivation 55 layer reducing a pore size of the passivation layer; and depositing a high refractive index material onto the sintered material wherein the high refractive index layer is uniform and crack free.

In another aspect, there is disclosed a multi-layered pig- 60 ment that includes a metal core. A passivation layer is applied to and surrounds the metal core. The passivation layer has a thickness less than or equal to 100 nm and a pore size distribution of 2.3 nm for 90 percent of the layer. A high refractive index layer is applied to and surrounds the passivation layer 65 wherein the high refractive index layer is uniform and crack

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In a further aspect, there is disclosed a multi-layered pigment that includes a metal core. A passivation layer is applied to and surrounds the metal core. The passivation layer has a thickness less than or equal to 100 nm and a pore size distribution of 2.3 nm for 90 percent of the layer. A high refractive index layer is applied to and surrounds the passivation layer wherein the high refractive index layer is uniform and crack free and has a thickness of from 50-200 nm.

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is a diagram detailing the operation pH range for a process for producing multilayer pigments;

FIG. 2 is graph of the pore size distribution of Al—SiO₂ material before and after densifying;

FIG. 3 are SEM images of the metal core and SiO₂ layer before and after densifying;

FIG. 4a is an SEM image of a pigment including TiO2

FIG. 4b is an EDX image of the pigment of FIG. 3a detailing the formation of the TiO₂ layer over the SiO₂—Al par-

FIG. 4c is an SEM images of a TiO₂ coated SiO₂—Al

DETAILED DESCRIPTION OF THE PREFERRED **EMBODIMENTS**

There is disclosed a process of forming a multi-layered pigment and a multi layered pigment that has a uniform surface coating for use in high chroma and other pigment systems. The process provides a low cost sol-gel process using an aqueous system to produce pigments in an economical manner. The multi-layered pigment includes a uniform coating of a high refractive index material that does not include cracks or other imperfections.

The process of the present invention allows for an economical procedure to produce pigments over a broad pH range and eliminates side reactions and processing problems in prior art applications. The process solves the problems of both etching of the Al core and a change in pH near the SiO₂—Al surface which are undesired for TiO₂ deposition. The process limits diffusion of protons through the SiO₂ layer that may cause a reaction with Al such that hydrogen gas is released causing weak adhesion of the TiO₂ particles and the formation of channels or cracks in the SiO₂ and TiO₂ layers.

The process of forming a multi layered pigment includes the steps of: providing a metal core material; treating the 50 metal core material with an acid, depositing a passivation layer onto the metal core material; densifying the metal core material having the passivation layer reducing a pore size of the passivation layer; and depositing a high refractive index material onto the densified material wherein the high refractive index layer is uniform and crack free.

The metal core material may include various metals including Al, Cr and coated Al such as Al coated with a thin layer of SiO₂ which requires a further thickness. In one aspect, the passivation layer includes SiO₂ and the high refractive index layer includes TiO2. Other passivation layer materials may include Al₂O₃. Additional high refractive index materials may include Fe₂O₃, ZrO₂, Cr₂O₃, and Fe₃O₄. In one aspect, the passivation layer may have a thickness less than or equal to 100 nm. The high refractive index layer may have a thickness of from 50-200 nm.

The step of treating the metal core material may include suspending the metal core material in a solution of phospho3

ric acid and acetone and stirring for a specified time. The resulting particles may then be washed and filtered.

The step of depositing a passivation layer onto the metal core material may include suspending the treated metal core particles in ethanol and heating the suspension with the addition of water and ammonia and adding a solution of tetraethoxysilane while stirring. The resulting particles may then be washed and filtered.

After washing and filtering the particles may be densified such that the passivation layer has a pore size distribution of 10 2.3 nm for 90 percent of the layer. In one aspect, the particles may be densified at a temperature of less than or equal to 400° C. for a specified time.

The step of depositing the high refractive index material onto the densified material may include suspending the densified metal core material having the passivation layer in water and heating the suspension while adding titanium tetrachloride while maintaining the suspension at a pH of 1.8 for a specified time such that ${\rm TiO_2}$ or another high refractive index material is uniformly deposited on the densified particle without the formation of cracks or other imperfections. The densified particles limit diffusion of protons of the aqueous solution at the very low pH to avoid reaction with the metal core material.

In another aspect, there is disclosed a multi-layered pigment that includes a metal core. A passivation layer is applied to and surrounds the metal core. The passivation layer has a thickness less than or equal to 100 nm and a pore size distribution of 2.3 nm for 90 percent of the layer. A high refractive index layer is applied to and surrounds the passivation layer wherein the high refractive index layer is uniform and crack free.

In a further aspect, there is disclosed a multi-layered pigment that includes a metal core. A passivation layer is applied to and surrounds the metal core. The passivation layer has a thickness less than or equal to 100 nm and a pore size distribution of 2.3 nm for 90 percent of the layer. A high refractive index layer is applied to and surrounds the passivation layer wherein the high refractive index layer is uniform and crack free and has a thickness of from 50-200 nm.

EXAMPLES

Materials

 ${
m SiO_2}$ -coated Aluminum flakes were provided by Silberline 45 Manufacturing Co. Tetraethoxysilane (TEOS), Ethanol (99%), acetone, phosphoric acid (85%), aqueous ammonia solution (28%), and titanium (IV) chloride were purchased from Sigma-Aldrich Chemical Co. (St. Louis, Mo.).

Preparation of Anti-Corrosive Aluminum Pigment by 50 Increasing Thickness of SiO₂ layer:

The preparation of anti-corrosive aluminum pigment is as follows:

a) Treatment with a phosphoric acid compound:

0.11~g of phosphoric acid (85%) is added to a suspension in 55~ which 5~g of SiO_2 -coated aluminum flakes (thickness of SiO_2 is about 15~nm) are dispersed in 55~ml of acetone, and stirred for 30 minutes at room temperature. The suspension is then filtered and washed with acetone two times. The solid parts are filtered and the SiO_2 -coated aluminum flakes treated with 60~phosphoric acid are obtained.

b) Silica Deposition:

In a round bottom flask equipped with a reflux condenser, the aluminum flakes, treated with phosphoric acid compound, obtained in step a) are suspended in 85 ml of ethanol. The 65 suspension is heated to 65° C. after 22 g of water and 2.2 g of a 28% aqueous ammonia solution had been added. Next, a

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solution of 6.5 g of tetraethoxysilane diluted with 8 ml of ethanol was added to it in small amounts while stirring. This reaction mixture was stirred for 18 hours at 65° C., and then filtered from the liquid.

Densifying the SiO₂-Coated Aluminum Flakes

Anti-corrosive aluminum flakes are washed with water and then dried at room temperature for overnight. Then the flakes are heated in the furnace. The ramp is set as 10° C./min and the temperature is set at 400° C. with dwelling time for 1 hour for further layer deposition.

Depositing ${\rm TiO}_2$ on Anti-Corrosive Aluminum Flakes by Aqueous System (Water):

5 g of the anti-corrosive aluminum flakes are suspended in 33 ml of water and heated to 75 $^{\circ}$ C. under stirring. 3 ml of titanium tetrachloride solution (40 wt %) is dropped into the suspension at the flow rate of 0.01 ml/min while keeping the pH at 1.8 using sodium hydroxide of 32 wt %. When the reaction is finished, the solid parts are filtered, washed with water, and then dried at 100 $^{\circ}$ C.

The deposition of high refractive index or dielectric materials in the water over platelet or metal cores often requires a very low pH, such as pH 1.8. At such low pH, although passivation layers of SiO₂ have been applied at thicknesses of from ~50-100 nm, the protons in the aqueous solution can still diffuse through the SiO₂ layer and react with Al core during the long deposition period, such as four hours. This phenomenon is shown by an increase in the pH values of from 1.8 to 3.0 after one hour when TiO₂ is deposited onto meal core particles that have SiO₂ layers that have not been densified. This side reaction between the proton and Al as well as the resultant pH increase at the core surface renders the deposition of TiO₂ using prior art procedures unsatisfactory.

A process for producing multilayer pigment particles should be stable and produce uniform crack free particles over a wide pH range. Referring to FIG. 1, there is shown a diagram depicting the operational pH range for non densified Al core flakes (top), densified Al core flake (bottom), and various high refractive index or dielectric layer materials 40 (above). The overlap in operational pH range for both core and layer materials is needed for successful layer deposition over the selected core.

To decrease diffusion of aqueous solution in the SiO_2 layer, a thicker protection SiO_2 layer is desired. However, the color strength of pigments would be impaired if the low refractive index SiO_2 layer thickness is more than 100 nm.

The densification of the core metal particle with the SiO₂ layer prior to TiO₂ deposition reduces the pore size of the material and provides a narrow pore size distribution that limits the diffusion of protons to the Al core material. Referring to FIG. 2, there is shown a plot of the pore size distribution for both densified and untreated particles. As can be seen in the figure the untreated particles had a pore size of 10.1 for 90% of the particles while the densified particles had a pore size of 2.3 for 90% of the particles. The results in FIG. 2 clearly demonstrated the pore size distribution of SiO2 is reduced as there is a peak shift from large values to smaller ones. The additional SiO₂ layer was deposited uniformly over the metal core and has a smooth surface with a thickness of approximately 100 nm. Referring to FIG. 3, there are shown SEM images of the densified and untreated or original flake or particle confirming the uniform deposition of the SiO₂ layer.

The narrow pore size distribution in the SiO₂ layer results in uniform TiO₂ layer deposition over the SiO₂—Al surface with no cracks or defects. Referring to FIGS. 4a-c, there are shown SEM images of both the densified particle and an untreated particle. The densified particles as shown in FIGS.

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4a and 4b have a smooth uniform surface while the untreated particles of FIG. 4c have a granular rough surface.

The above examples and embodiments are for illustrative purposes only and changes, modifications, and the like will be apparent to those skilled in the art and yet still fall within the 5 scope of the invention. As such, the scope of the invention is defined by the claims.

We claim:

1. A process of forming a multi-layered pigment comprising the steps of:

providing a metal core material;

treating the metal core material with an acid,

depositing a passivation layer onto the metal core material; densifying the metal core material having the passivation layer forming a densified material and reducing a pore 15 size of the passivation layer;

- depositing a high refractive index material onto the densified material wherein the high refractive index layer is uniform and crack free wherein the passivation layer has a pore size distribution of 2.3 nm for 90 percent of the layer.
- 2. The process of forming the multi-layered pigment of claim 1 wherein the passivation layer includes SiO₂.
- 3. The process of forming the multi-layered pigment of claim 1 wherein the high refractive index layer includes TiO₂. 25
- **4**. The process of forming the multi-layered pigment of claim **1** wherein the metal or alloy core is selected from the group consisting of: Al, Cr, coated Al, Ag, Cu, brass, Mn.
- 5. The process of forming the multi-layered pigment of claim 1 wherein the high refractive index layer has a thickness 30 of from 50-200 nm.
- **6**. The process of forming the multi-layered pigment of claim **1** wherein the passivation layer has a thickness less than or equal to 100 nm.
- 7. The process of forming the multi-layered pigment of 35 claim 1 wherein the step of treating includes suspending the metal core material in a solution of phosphoric acid and acetone and stirring for a specified time.
- **8**. The process of forming the multi-layered pigment of claim **7** wherein the step of treating further includes the step 40 of washing and filtering the metal core material.
- 9. The process of forming the multi-layered pigment of claim 1 wherein the step of depositing the passivation layer onto the metal core material includes suspending the treated metal core particles in ethanol and heating the suspension 45 with the addition of water and ammonia and adding a solution of tetraethoxysilane while stirring.

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- 10. The process of forming the multi-layered pigment of claim 9 wherein the step of depositing the passivation layer onto the metal core material further includes the step of washing and filtering.
- 11. The process of forming the multi-layered pigment of claim 1 wherein the passivation layer is densified at a temperature of less than or equal to 400° C. for a specified time.
- 12. The process of forming the multi-layered pigment of claim 1 wherein the step of depositing the high refractive index material onto the densified material includes suspending the densified metal core material having the passivation layer in water and heating the suspension while adding titanium tetrachloride while maintaining the suspension at a pH of 1.8.
 - **13**. A multi-layered pigment comprising: a metal core;
 - a passivation layer applied to and surrounding the metal core, the passivation layer having a thickness less than or equal to 100 nm and a pore size distribution of 2.3 nm for 90 percent of the layer;
 - a high refractive index layer applied to and surrounding the passivation layer wherein the high refractive index layer is uniform and crack free.
- 14. The multi-layered pigment of claim 13 wherein the passivation layer includes ${\rm SiO}_2.$
- 15. The multi-layered pigment of claim 13 wherein the high refractive index layer includes TiO₂.
- 16. The multi-layered pigment of claim 13 wherein the passivation layer is densified prior to application of the high refractive index layer.
- 17. The multi-layered pigment of claim 13 wherein the metal core or alloy core is selected from the group consisting of: Al, Cr, coated Al, Ag, Cu, Mn, brass.
- **18**. The multi-layered pigment of claim **13** wherein the high refractive index layer has a thickness of from 50-200 nm.
 - **19**. A multi-layered pigment comprising: a metal core;
 - a passivation layer applied to and surrounding the metal core, the passivation layer having a thickness less than or equal to 100 nm and a pore size distribution of 2.3 nm for 90 percent of the layer;
 - a high refractive index layer applied to and surrounding the passivation layer wherein the high refractive index layer is uniform and crack free and has a thickness of from 50-200 nm.

* * * * *

UNITED STATES PATENT AND TRADEMARK OFFICE **CERTIFICATE OF CORRECTION**

PATENT NO. : 9,359,504 B2

APPLICATION NO. : 14/323085

DATED : June 7, 2016

INVENTOR(S) : Songtao Wu et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

On the Title Page, Item (54) and in the Specification at Column 1, line 3, in Title, delete "AL" and insert --Al--.

Signed and Sealed this Sixth Day of September, 2016

Michelle K. Lee

Michelle K. Lee

Director of the United States Patent and Trademark Office